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JEE Advanced - 2022 Paper-1

(Held on 28th August, 2022)

QUESTIONS & SOLUTIONS

PART A: PHYSICS

SECTION-1: (Maximum Marks: 24)

- This section contains Eight (08) question stems.
- The answer to each question is a **NUMERICAL VALUE**.
- For each question, enter the correct numerical value of the answer using the mouse and the
- on-screen virtual numeric keypad in the place designated to enter the answer. If the numerical
- value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 **ONLY** if the correct numerical value is entered at the designated place;

Zero Marks : 0 In all other cases.

1. Two spherical stars A and B have densities ρ_A and ρ_B , respectively. A and B have the same radius, and their masses M_A and M_B are related by $M_B = 2M_A$. Due to an interaction process, star A loses some of its mass, so that its radius is halved, while its spherical shape is retained, and its density remains ρ_A . The entire mass lost by A is deposited as a thick spherical shell on B with the density of the shell being ρ_A . If ν_A and ν_B are the escape velocities from A and B after the interaction process, the ratio

$$\frac{\nu_{B}}{\nu_{A}} = \sqrt{\frac{10n}{15^{1/3}}} \ . \ The \ value \ of \ n \ is \underline{\hspace{2cm}}$$

Ans. 2.30

Sol. Given $RA = R_B = R$

$$M_B = 2M_A$$

Calculation of escape velocity for A:

Radius of remaining star = $\frac{R_A}{2}$.

Mass of remaining star = $\rho_A \frac{4}{3} \pi \frac{R_A^3}{8} = \frac{M_A}{8}$

$$\frac{-GM_{_{A/B}}}{R_{_{A/2}}} + \frac{1}{2}mv_{_{A}}^2 = 0 \Rightarrow v_{_{A}} = \sqrt{\frac{2GMA_{_{B}}}{R_{_{A/2}}}} = \sqrt{\frac{GM_{_{A}}}{2R}}$$

Calculation of escape velocity for B

Mass collected over $B = \frac{7}{8}M_A$

Let the radius of B becomes r.

$$\therefore \frac{4}{3} \pi \Big(r^3 - R_{\scriptscriptstyle B}^3 \Big) \rho_{\scriptscriptstyle A} = \frac{7}{8} \rho_{\scriptscriptstyle A} \frac{4}{3} \pi R_{\scriptscriptstyle A}^3 \\ \Rightarrow \pi^3 = \frac{7}{8} R_{\scriptscriptstyle A}^3 + R_{\scriptscriptstyle B}^3 = \frac{(15)^{1/3} R}{2}$$

$$\therefore \frac{V_{\text{B}}^2}{2} = \frac{23GM_{\text{A}}}{8 \times 15^{1/3} \frac{R}{2}} = \frac{23GM_{\text{A}}}{4 \times 15^{1/3} R}$$

$$\therefore V_B = \sqrt{\frac{23GM_A}{2 \times 15^{1/3}R}}$$

$$\therefore \frac{V_B}{V_*} = \sqrt{\frac{23}{15^{1/3}}} = \sqrt{\frac{10 \times 2.30}{15^{1/3}}}$$

n = 2.30

2. The minimum kinetic energy needed by an alpha particle to cause the nuclear reaction ${}^{16}_{7}\text{N} + {}^{4}_{2}\text{He} \rightarrow {}^{1}_{1}\text{H} + {}^{19}_{8}\text{O}$ in a laboratory frame is n (in MeV). Assume that ${}^{16}_{7}\text{N}$ is at rest in the laboratory frame. The masses of ${}^{16}_{7}\text{N}, {}^{4}_{2}\text{He}, {}^{1}_{1}\text{H}$ and ${}^{19}_{8}\text{O}$ can be taken to be 16.006 u, 4.003 u, 1.008 u and 19.003 u, respectively, where 1 u = 930 MeVc⁻². The value of n is ______.

Ans. 2.32 to 2.33

Sol.
$${}^{16}_{7}\text{N} + {}^{4}_{2}\text{He} \rightarrow {}^{1}_{1}\text{He} + {}^{19}_{8}\text{O}$$

$$^{16}_{7}\text{N}$$
 + $^{4}_{2}\text{He}$ \rightarrow $^{1}_{1}\text{He}$ + $^{19}_{8}\text{C}$

$$4v_0 = 1v_1 + 19v_2 = 20v_2$$

(For max loss of KE)

$$V_0 = \frac{V_2}{5}$$

E required = $(1.008 + 19.003 - 16.006 - 4.003) \times 930 = 1.86$

$$\frac{1}{2}4v_0^2 - \frac{1}{2}20v^2 = 1.86$$

$$\frac{1}{2}4v_0^2 - 10\frac{v_0^2}{25}20v^2 = 1.86$$

$$2v_0^2 - \frac{2}{5}v_0^2 = 1.86$$

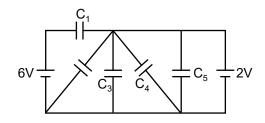
$$\frac{8}{5}v_0^2 = 1.86$$

$$V_0^2 = \frac{1.86 \times 5}{8}$$

$$KE = \frac{1}{2}4v_0^2 = 2v_0^2 = \frac{18.6 \times 5}{4}$$

$$= 2.325$$

3. In the following circuit C_1 = 12 μ F, C_2 = C_3 = 4 μ F and C_4 = C_5 = 2 μ F. The Charge stored in C_3 is μ C.

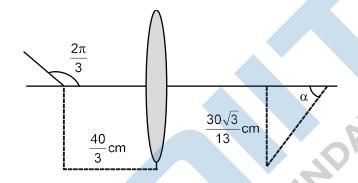


Ans.

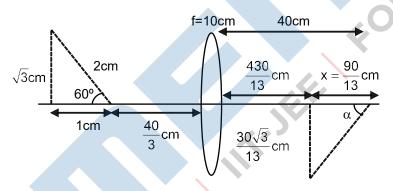
Sol. Potential difference across the terminals of C_3 is 2V.

$$Q_3 = CV = (4\mu)(2) = 8\mu C$$

4. A rod of length 2 cm makes an angle $\frac{2\pi}{3}$ rad with the principal axis of a thin convex lens. The lens has a focal length of 10 cm and is placed at a distance of $\frac{40}{3}$ cm from the object as shown in the figure. The height of the image is $\frac{30\sqrt{3}}{13}$ cm and the angle made by it with respect to the principal axis is α rad. The value of α is $\frac{\pi}{n}$ rad, where n is ______.



Ans. 6



Sol.

$$\frac{h_{_{i}}}{h_{_{0}}} = \frac{v}{u} \Rightarrow \frac{-\frac{30\sqrt{3}}{13}}{\sqrt{3}} = \frac{v}{\frac{43}{3}} \Rightarrow v_{_{1}} = \frac{430}{13} cm$$

$$*\frac{1}{V} - \frac{1}{U} = \frac{1}{f} \Rightarrow \frac{1}{V} = \frac{1}{10} - \frac{3}{40} \Rightarrow V = 40cm$$

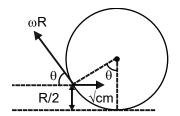
$$*x = 40 - \frac{430}{13} = \frac{90}{13}$$
cm

$$tan\alpha = \frac{\frac{30\sqrt{3}}{13}}{\frac{90}{13}} = \frac{1}{\sqrt{3}} \Rightarrow \alpha = 30^{\circ} = \frac{\pi}{6}$$

N = 6 Ans.

5. At time t = 0, a disk of radius 1m starts to roll without slipping on a horizontal plane with an angular acceleration of $\alpha = \frac{2}{3}$ rad s⁻². A small stone is stuck to the disk. At t = 0, it is at the contact point of the disk. The maximum height (in m) reached by the stone measured from the plane is $\frac{1}{2} + \frac{x}{10}$. The value of x is _____. [Take $g = 10 \text{ ms}^{-2}$.]

Ans. 0.52



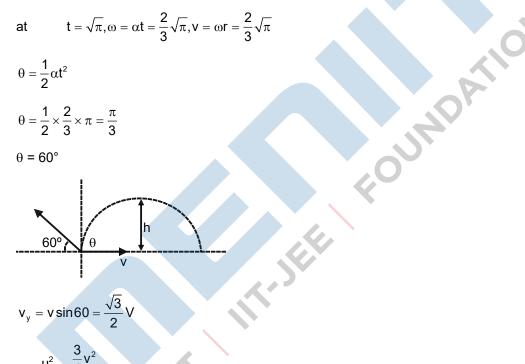
Sol.

 $t = 0, \omega = 0$ Αt

at
$$t = \sqrt{\pi}, \omega = \alpha t = \frac{2}{3}\sqrt{\pi}, v = \omega r = \frac{2}{3}\sqrt{\pi}$$

$$\theta = \frac{1}{2}\alpha t^2$$

$$\theta = \frac{1}{2} \times \frac{2}{3} \times \pi = \frac{\pi}{3}$$



$$v_y = v \sin 60 = \frac{\sqrt{3}}{2}V$$

$$h = \frac{u_y^2}{2a} = \frac{\frac{3}{4}v^2}{2a}$$

$$h = \frac{\frac{3}{4} \times \frac{4}{9}\pi}{2\alpha}$$

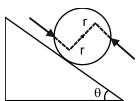
$$h=\frac{3\pi}{9\times 2g}=\frac{\pi}{6g}$$

Maximum height from plane, $H = \frac{R}{2} + h$

$$H = \frac{1}{2} + \frac{\pi}{6 \times 10}$$

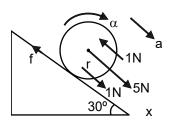
$$x = \frac{\pi}{6}$$
; $x = 0.52$

6. A solid sphere of mass 1 kg and radius 1 m rolls without slipping on a fixed inclined plane with an angle of inclination $\theta = 30^{\circ}$ from horizontal. Two forces of magnitude 1 N each, parallel to the incline, act on the sphere, both at distance r = 0.5 m from the center of the sphere, as shown in the figure. The acceleration of the sphere down the plane is ______ ms⁻². (Take g = 10 ms⁻².)



Ans. 2.85 to 2.86

Sol. Solid sphere 1kg, 1m



$$5 + 1 - 1 - f = 1a$$

$$5 - f = a$$

About COM

$$f1 - 2(1(0.5)) = \frac{2}{5}Mr^2\alpha$$

$$\Rightarrow$$
 f - 1 = $\frac{2}{5}$ a \Rightarrow f = 1 + $\frac{2}{5}$ a

$$5 - a = 1 + \frac{2}{5}a$$

$$\Rightarrow 4 = \frac{7a}{5} \Rightarrow a = \frac{20}{7} = 2.86 \text{m/s}^2$$

Consider an LC circuit, with inductance L = 0.1 H and capacitance C = 10^{-3} F, kept on a plane. The area of the circuit is 1 m². It is placed in a constant magnetic field of strength B₀ which is perpendicular to the plane of the circuit. At time t = 0, the magnetic field strength starts increasing linearly as B = B₀ + β t with β = 0.04 Ts⁻¹. The maximum magnitude of the current in the circuit is ____ mA.

Ans.

Sol. Maximum energy will be

$$\frac{q_0^2}{2C} = \frac{1}{2}LI_0^2$$

$$\frac{q_0^2}{CL} = I_0^2$$

$$\boldsymbol{I_0} = \frac{\boldsymbol{q_0}}{\sqrt{LC}}$$

$$I_0 = \frac{CV}{\sqrt{LC}}$$

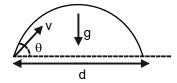
$$I_0 = \sqrt{\frac{C}{L}} \times V$$
 $V = emf = \left| \frac{AdB}{dt} \right|$

$$I_0 = \sqrt{\frac{10^{-3}}{0.1}} \times 0.04$$
 $V = (1 \times 0.04)$

Maximum current $I_0 = 0.004 = 4mA$

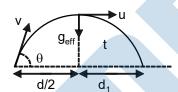
8. A projectile is fired from horizontal ground with speed v and projection angle θ . When the acceleration due to gravity is g, the range of the projectile is d. If at the highest point in its trajectory, the projectile enters a different region where the effective acceleration due to gravity is $g' = \frac{g}{0.81}$, then the new range is d' = nd. The value of n is _____.

Ans. 0.95



Sol.

$$d = \frac{v^2 \sin 2\theta}{g}$$



$$H_{\text{max}} = \frac{v^2 \sin^2 \theta}{2g}; \frac{1}{2} g_{\text{eff}} t^2 = H_{\text{max}} \Rightarrow t^2 = \frac{2 H_{\text{max}}}{g_{\text{eff}}}; t = \sqrt{\frac{v^2 \sin^2 \theta \times 0.81}{g^2}}; t = \frac{0.9 v \sin \theta}{g}$$

$$t^2 = \frac{2 \times v^2 \sin^2 \theta}{2g \left(\frac{g}{0.81}\right)}$$

d' = New range =
$$\frac{d}{2} + d_1$$

 $d_1 = v \cos \theta^{o} t$

$$=\frac{v^2 \, sin^2 \, \theta \cos \theta \times 0.9}{g}; d^{'}=\frac{v^2 \, sin \, 2\theta}{2g}+\frac{v^2 \, sin \, 2\theta \times 0.9}{2g}$$

$$=\frac{v^2\sin 2\theta}{q}\left(\frac{1.0}{2}\right)=0.95d$$

n = 0.95

SECTION-2: (Maximum Marks: 24)

This section contains SIX (06) questions.

Each question has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is (are) correct answer(s).

For each question, choose the option(s) corresponding to (all) the correct answer(s).

Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** (all) the correct option(s) is(are) chosen;

Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

: +2 If three or more options are correct but **ONLY** two options are chosen, Partial Marks

both of which are correct:

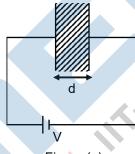
: +1 If two or more options are correct but **ONLY** one option is chosen and it Partial Marks

is a correct option;

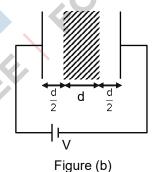
Zero Marks : 0 If unanswered;

: -2 In all other cases. Negative Marks

9. A medium having dielectric constant K > 1 fills the space between the plates of a parallel plate capacitor. The plates have large area, and the distance between them is d. The capacitor is connected to a battery of voltage V, as shown in figure (a). Now, both the plates are moved by a distance of $\frac{d}{2}$ from their original positions, as shown in figure (b).





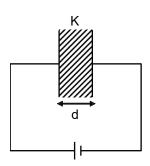


In the process of going from the configuration depicted in Figure (a) to that in Figure (b), which of the following statement(s) is(are) correct?

- (A) The electric field inside the dielectric material is reduced by a factor of 2K.
- (B) The capacitance is decreased by a factor of $\frac{1}{k'+1}$.
- (C) The voltage between the capacitor plates is increased by a factor of (K + 1).
- (D) The work done in the process **DOES NOT** depend on the presence of the dielectric material.

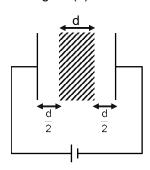
Ans. (B)

Sol. For figure (a)



$$\boldsymbol{E}_0 = \frac{\boldsymbol{V}}{\boldsymbol{d}}; \boldsymbol{C} = \frac{\boldsymbol{K}\boldsymbol{\epsilon}_0 \boldsymbol{A}}{\boldsymbol{d}}$$

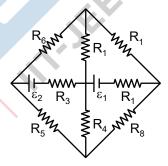
For figure (b)



$$C^{'} = \frac{\epsilon_0 A}{2d - d + d / k}$$

$$C^{'} = \frac{K\epsilon_0 A}{(K+1)d}; C^{'} = \frac{C}{K+1}$$

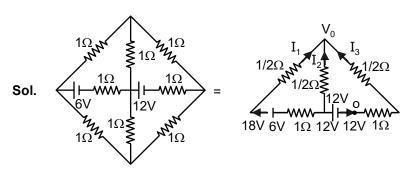
10. The figure shows a circuit having eight resistances of 1 Ω each, labelled R₁ to R₈, and two ideal batteries with voltages ε_1 = 12 V and ε_2 = 6 V.



Which of the following statement(s) is(are) correct?

- (A) The magnitude of current flowing through R_1 is 7.2 A.
- (B) The magnitude of current flowing through R_2 is 1.2 A.
- (C) The magnitude of current flowing through R_3 is 4.8 A.
- (D) The magnitude of current flowing through $R_{\rm 5}$ is 2.4 A.

Ans. (A,B,C,D)



From KCL
$$i_{1} + i_{2} + i_{3} = 0$$

$$\Rightarrow \frac{18 - V_{0}}{3/2} + \frac{12 - V_{0}}{1/2} + \frac{0 - V_{0}}{3/2} = 0$$

$$\Rightarrow 18 - V_{0} + 36 - 3V_{0} - V_{0} = 0$$

$$\Rightarrow 54 = 5V_{0}$$

$$\frac{2\left(\frac{54}{5} - v'\right)}{1} + \frac{18 - v'}{1} = 0$$

$$\Rightarrow \frac{108}{5} + 18 = 3V'$$

$$\Rightarrow v' = \frac{198}{5 \times 3} = \frac{66}{5}V$$

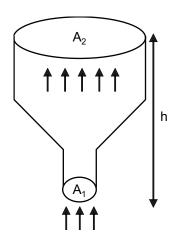
$$I_{R_{1}} = \frac{36}{5} = 7.2A$$

$$I_{R_{2}} = \frac{6}{5} = 1.2A$$

$$I_{R_{3}} = \frac{24}{5} = 4.8A$$

 $I_{R_5} = \frac{12}{5} = 2.4A$

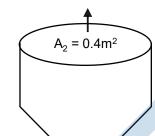
An ideal gas of density ρ = 0.2 kg m⁻³ enters a chimney of height h at the rate of α = 0.8 kg s⁻¹ from its lower end, and escapes through the upper end as shown in the figure. The cross-sectional area of the lower end is A₁ = 0.1 m² and the upper end is A₂ = 0.4 m². The pressure and the temperature of the gas at the lower end are 600 Pa and 300 K, respectively, while its temperature at the upper end is 150 K. The chimney is heat insulated so that the gas undergoes adiabatic expansion. Take g = 10 ms⁻² and the ratio of specific heats of the gas γ = 2. Ignore atmospheric pressure.



Which of the following statement(s) is(are) correct?

- (A) The pressure of the gas at the upper end of the chimney is 300 Pa.
- (B) The velocity of the gas at the lower end of the chimney is 40 ms⁻¹ and at the upper end is 20 ms⁻¹.
- (C) The height of the chimney is 590 m.
- (D) The density of the gas at the upper end is 0.05 kg m^{-3} .

Ans. (B)



Sol.

$$A_1 = 0.1 \text{m}^2$$
 $P_1 = 600 \text{ Pa}$
 $T_1 = 300 \text{ K}$

$$\frac{dm}{dt} = \rho_1 A_1 V_1 = 0.8 kg / sA$$

$$v_1 = \frac{0.8}{0.2 \times 0.1} = 40 \text{m/s}$$

$$g = 10 \text{ m/s}^2$$

$$\gamma = 2$$

Gas undergoes adiabatic expansion,

$$p^{1-\gamma} T^{\gamma} = Constant$$

$$\frac{P_2}{P_1} = \left(\frac{T_1}{T_2}\right)^{\frac{r}{1-y}}$$

$$P_2 = \left(\frac{300}{150}\right)^{\frac{2}{-1}} \times 600$$

$$P_2 = \frac{600}{4} = 150 Pa$$

Now
$$\rho = \frac{PM}{RT} \Rightarrow \rho \propto \frac{P}{T}$$

$$\frac{\rho_1}{\rho_2} = \left(\frac{P_1}{P_2}\right) \left(\frac{T_1}{T_2}\right) = \left(\frac{150}{600}\right) \left(\frac{300}{150}\right) = \frac{1}{2}$$

$$\rho_2 = \frac{\rho_1}{2} = 0.1 kg / m^3$$

Now
$$\rho_2 A_2 V_2 = 0.8 \Rightarrow V_2 = \frac{0.8}{0.1 \times 0.4} = 20 \text{m/s}$$

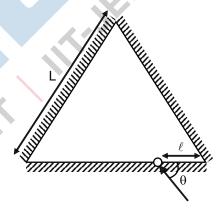
$$P_{1}A_{1}\Delta x_{1}-P_{2}A_{2}\Delta x_{2}=\frac{1}{2}\Delta mVV_{2}^{2}-\frac{1}{2}\Delta mVV_{1}^{2}+\Delta mgh+\frac{f}{2}\big(P_{2}\Delta V_{2}-P_{1}\Delta V_{1}\big)$$

$$\Rightarrow 2P_1\frac{\Delta V_1}{\Delta m}-2P_2\frac{\Delta V_2}{\Delta m}=\frac{V_2^2-V_1^2}{2}+gh$$

$$\Rightarrow \frac{2 \times 600}{0.2} - \frac{2 \times 150}{0.1} = \frac{20^2 - 40^2}{2} + 10h$$

$$h = 360 \text{ m}$$

12. Three plane mirrors form an equilateral triangle with each side of length L. There is a small hole at a distance $\ell > 0$ from one of the corners as shown in the figure. A ray of light is passed through the hole at an angle θ and can only come out through the same hole. The cross section of the mirror configuration and the ray of light lie on the same plane.



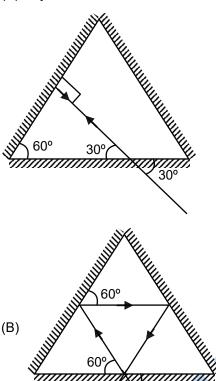
Which of the following statement(s) is(are) correct?

- (A) The ray of light will come out for θ = 30°, for 0 < ℓ < L.
- (B) There is an angle for $\ell = \frac{L}{2}$ at which the ray of light will come out after two reflections.
- (C) The ray of light will **NEVER** come out for θ = 60°, and $\ell = \frac{L}{3}$.

(D) The ray of light will come out for θ = 60°, and $0<\ell<\frac{L}{2}$ after six reflections.

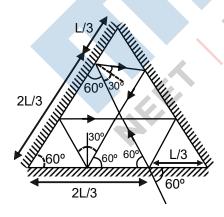
Ans. (A, B)

(A) Ray will come out after one reflection for θ = 30° & 0 < ℓ < L Sol.

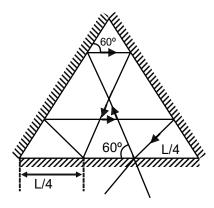


For θ = 60° & $\ell = \frac{L}{2}$, ray will come out after two reflections.

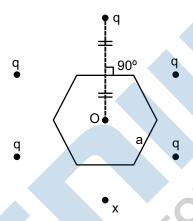
(C) For $\,\ell=\frac{L}{3}\,$ & θ = 60° ray will come out after five reflections.



(D) for θ = 60° & 0 < ℓ < $\frac{L}{2}$, ray will come out after five reflection.



13. Six charges are placed around a regular hexagon of side length a as shown in the figure. Five of them have charge q, and the remaining one has charge x. The perpendicular from each charge to the nearest hexagon side passes through the center O of the hexagon and is bisected by the side.

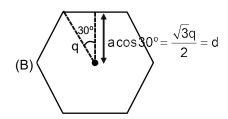


Which of the following statement(s) is(are) correct in SI units?

- (A) When x = q, the magnitude of the electric field at O is zero.
- (B) When x = q, the magnitude of the electric field at O is $\frac{q}{6\pi \in_0 a^2}$.
- (C) When x = 2q, the potential at O is $\frac{7q}{4\sqrt{3}\pi \in_0 a}$.
- (D) When x = –3q, the potential at O is $\frac{3q}{4\sqrt{3}\pi} \in_0 a$.

Ans. (A,B,C)

Sol. (A) Due to symmetry $\overrightarrow{E}_0 = 0$



$$\mathsf{E}_{\mathsf{net}} \, = \frac{kq}{\left(2\mathsf{d}\right)^2} \times 2 = \frac{2q \times 4}{4\pi\epsilon_0 \cdot 4 \cdot 3a^2}$$

$$=\frac{q}{6\pi\epsilon_0 a^2}$$

(C)
$$v = \frac{7kq}{2d} = \frac{7q}{4\pi\epsilon_0 \cdot \sqrt{3a}} = \frac{7q}{4\sqrt{3}\pi\epsilon_0 q}$$

(D)
$$v = \frac{2kq}{2d} = \frac{2q}{4\pi\epsilon_0 \cdot \sqrt{3}a} = \frac{q}{2\sqrt{3}\pi\epsilon_0 q}$$

14. The binding energy of nucleons in a nucleus can be affected by the pairwise Coulomb repulsion. Assume that all nucleons are uniformly distributed inside the nucleus. Let the binding energy of a proton be E_b^p and the binding energy of a neutron be E_bⁿ in the nucleus.

Which of the following statement(s) is(are) correct?

- (A) $E_h^p E_h^n$ is proportional to Z(Z-1) where Z is the atomic number of the nucleus.
- (B) $E_b^p E_b^n$ is proportional to $A^{-\frac{1}{3}}$ where A is the mass number of the nucleus.
- (C) $E_b^p E_b^n$ is positive.
- (D) E_b increases if the nucleus undergoes a beta decay emitting a positron.

Ans. (A,B,D)

Sol. Binding energy of proton & neutron due to nuclear force is same. So difference in binding energy is only due to electrostatic P.E. and it is positive

$$E_0^P - E_0^n$$
 = electrostatic P.E.

$$=Z\times\frac{1}{4\pi\epsilon_0}\frac{(Z-1)e^2}{R}$$

Where
$$R = R_0 A^{1/3}$$

$$= \frac{1}{4\pi\epsilon_0} \frac{Z(Z-1)e^2}{R_0 A^{\frac{1}{3}}}$$

SECTION-3: (Maximum Marks: 12)

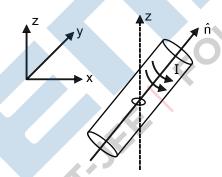
- This section contains **FOUR (04)** Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has TWO lists: List-I and List-II.
- List-I has Four entries (I), (II), (III) and (IV) and List-II has Five entries (P), (Q), (R), (S) and (T).
- FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE
 of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 ONLY if the option corresponding to the correct combination is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

15. A small circular loop of area A and resistance R is fixed on a horizontal xy-plane with the center of the loop always on the axis \hat{n} of a long solenoid. The solenoid has m turns per unit length and carries current I counterclockwise as shown in the figure. The magnetic field due to the solenoid is in \hat{n} direction. List-I gives time dependences of \hat{n} in terms of a constant angular frequency ω . List-II gives the torques experienced by the circular loop at time $t = \frac{\pi}{6\omega}$, Let $\alpha = \frac{A^2 \mu_0^2 m^2 l^2 \omega}{2R}$.



List I

(I) $\frac{1}{\sqrt{2}}(\sin\omega t \hat{j} + \cos\omega t \hat{k})$

(II)
$$\frac{1}{\sqrt{2}}(\sin\omega t \hat{i} + \cos\omega t \hat{j})$$

(III)
$$\frac{1}{\sqrt{2}} (\sin \omega t \hat{i} + \cos \omega t \hat{k})$$

(IV)
$$\frac{1}{\sqrt{2}}(\cos\omega t \hat{i} + \sin\omega t \hat{k})$$

(D)
$$-\frac{\alpha}{4}\hat{i}$$

(R)
$$\frac{3\alpha}{4}\hat{i}$$

(S)
$$\frac{\alpha}{4}\hat{j}$$

(T)
$$-\frac{3\alpha}{4}\hat{i}$$

Which one of the following options is correct?

(C) I-Q, II-P, III-S, IV-R

(D) I-T, II-Q, III-P, IV-R

Ans. (C)

Sol. (I)
$$\vec{B} = \frac{\mu_0 mI}{\sqrt{2}} (\sin \omega t \hat{j} + \cos \omega t \hat{k})$$

$$\phi = \overrightarrow{B} \cdot \overrightarrow{A} = \frac{\mu_0 mI}{\sqrt{2}} cos(\omega t) \cdot A$$

$$\varepsilon = \frac{d\phi}{dt} = \frac{\mu_0 m I \omega A}{\sqrt{2}} sin(\omega t)$$

$$i = \frac{\varepsilon}{R} = \frac{\mu_0 m I \omega A}{\sqrt{2R}} sin(\omega t)$$

$$\vec{M} = i\vec{A} = iA(\hat{k}) = \frac{\mu_0 m I_{\odot} A^2}{\sqrt{2}R} sin(\omega t)(\hat{k})$$

$$\vec{\tau} = \vec{M} \times \vec{B} = \frac{\mu_0 m^2 l^2 \omega A^2}{\sqrt{2} R} sin^2(\omega t)(-\hat{i})$$

$$= -\left(\frac{\alpha}{4}\right) \hat{i}$$
(II) $\vec{B} = \frac{\mu_0 m I}{\sqrt{2}} (sin\omega t \hat{i} + cos\omega t \hat{j})$

$$\phi = 0, \ \epsilon = 0, \ i = 0, \ t = 0$$
(III) $\vec{B} = \frac{\mu_0 m I}{\sqrt{2}} (sin\omega t \hat{i} + cos\omega t \hat{k})$

$$\phi = \vec{B} \cdot \vec{A} = \frac{\mu_0 m I}{\sqrt{2}} \cdot cos(\omega t) \cdot \vec{A}$$

$$\epsilon = -\frac{d\phi}{dt} = \frac{\mu_0 m I \omega A}{\sqrt{2}} sin(\omega t)$$

$$i = \frac{\epsilon}{R} = \frac{\mu_0 m I \omega A}{\sqrt{2} R} sin(\omega t)$$

$$=-\left(\frac{\alpha}{4}\right)\hat{i}$$

(II)
$$\overrightarrow{B} = \frac{\mu_0 mI}{\sqrt{2}} (\sin \omega t \hat{i} + \cos \omega t \hat{j})$$

$$\phi = 0, \ \epsilon = 0, \ i = 0, \ t = 0$$

(III)
$$\overrightarrow{B} = \frac{\mu_0 mI}{\sqrt{2}} (\sin \omega t \hat{i} + \cos \omega t \hat{k})$$

$$\phi = \vec{B} \cdot \vec{A} = \frac{\mu_0 m I}{\sqrt{2}} \cdot cos(\omega t) \cdot A$$

$$\varepsilon = -\frac{d\phi}{dt} = \frac{\mu_0 m I \omega A}{\sqrt{2}} sin(\omega t)$$

$$i = \frac{\varepsilon}{R} = \frac{\mu_0 m I \omega A}{\sqrt{2R}} sin(\omega t)$$

$$\overrightarrow{M} = \overrightarrow{iA} = iA(\widehat{k}) = \frac{\mu_0 m I \omega A^2}{\sqrt{2}R} sin(\omega t)(\widehat{k})$$

$$\vec{\tau} = \vec{M} \times \vec{B} = \frac{\mu_0 m^2 l^2 \omega A^2}{2R} \sin^2(\omega t) (+\hat{j})$$

$$=\frac{\alpha}{4}\hat{j}$$

(IV)
$$\vec{B} = \frac{\mu_0 mI}{\sqrt{2}} (\cos \omega t \hat{j} + \sin \omega t \hat{k})$$

$$\phi = \overrightarrow{B} \cdot \overrightarrow{A} = \frac{\mu_0 mI}{\sqrt{2}} \cdot \sin(\omega t) \cdot A$$



$$\epsilon = -\frac{d\phi}{dt} = \frac{\mu_0 m I \omega A}{\sqrt{2}} cos(\omega t)$$

$$i = \frac{\varepsilon}{R} = -\frac{\mu_0 m I_{\omega} A}{\sqrt{2} R} \cos(\omega t)$$

$$\vec{M} = i\vec{A} = iA(\hat{k}) = -\frac{\mu_0 m I_{\odot} A^2}{\sqrt{2}R} cos(\omega t)(\hat{k})$$

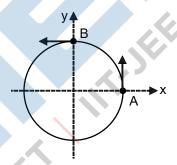
$$\overset{\rightarrow}{\tau} = \vec{M} \times \vec{B} = -\frac{\mu_0 m^2 l^2 \omega A^2}{2R} cos^2 (\omega t) (-\hat{i})$$

$$=\alpha\cdot\cos^2\left(\frac{\pi}{6}\right)\hat{\mathbf{i}}$$

$$=\frac{3\alpha}{4}\hat{i}$$

- 16. List I describes four systems, each with two particles A and B in relative motion as shown in figure. List II gives possible magnitudes of then relative velocities (in ms⁻¹) at time $t = \frac{\pi}{3}s$.
 - List I List II
 - (I) A and B are moving on a horizontal circle of radius (P) $\frac{\sqrt{3} + \sqrt{3}}{2}$

1 m with uniform angular speed ω = 1 rad s⁻¹. The initial angular positions of A and B at time t = 0 are θ = 0 and θ = $\frac{\pi}{2}$ respectively.

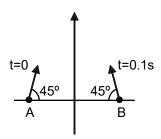


(II) Projectiles A and B are fired (in the same vertical plane) at t = 0 (Q) $\frac{(\sqrt{3}-1)}{\sqrt{2}}$

and t = 0.1 s respectively, with the same speed $\,\nu = \frac{5\pi}{\sqrt{2}}\,$ ms $^{-1}$ and

at 45° from the horizontal plane. The initial separation between A and

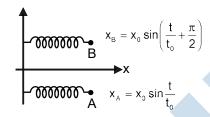
B is large enough so that they do not collide, $(g = 10 \text{ ms}^{-2})$.



- (III) Two harmonic oscillators A and B moving in the x direction
- (R) $\sqrt{10}$

according to
$$\mathbf{x}_{\mathrm{A}}$$
 = \mathbf{x}_{0} $\sin\frac{t}{t_{\mathrm{0}}}$ and $\mathbf{x}_{\mathrm{B}} = \mathbf{x}_{\mathrm{0}}\sin\left(\frac{t}{t_{\mathrm{0}}} + \frac{\pi}{2}\right)$ respective,

starting from t = 0. Take $x_0 = 1$ m, $t_0 = 1$ s.

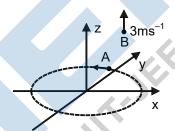


- (IV) Particle A is rotating in a horizontal circular path of radius
- (S) $\sqrt{2}$

1 m on the xy plane, with constant angular speed $\omega = 1 \text{ rad s}^{-1}$.

Particle B is moving up at a constant speed 3 m s⁻¹ in the vertical

Direction as shown in the figure. (Ignore gravity.)



Which one of the following options is correct?

Ans. (C)

Sol. (I)
$$V_{BA}^2 = V_A^2 + V_B^2 - 2V_{AB} \cos \theta$$

As $\omega_{A} = \omega_{B}$, $\theta = 90^{\circ}$ remains constant.

Also,
$$v_A = v_B = 1 \text{ m/s}$$

So,
$$v_{BA} = \sqrt{2} m / s$$

(II)
$$\vec{u}_A = \frac{5\pi}{2}\hat{i} + \frac{5\pi}{2}\hat{j}$$

$$\vec{V}_A = \frac{5\pi}{2} \mathbf{i} + + \left(\frac{5\pi}{2} - 10 \cdot \frac{\pi}{3} \right) \hat{\mathbf{j}}$$

$$=\frac{5\pi}{2}\hat{\mathbf{i}}-\frac{5\pi}{6}\hat{\mathbf{j}}$$

$$\vec{u}_{\text{B}} = -\frac{5\pi}{2}\hat{i} + \frac{5\pi}{2}\hat{j}$$

$$\vec{u}_{B} = -\frac{5\pi}{2}\hat{i} - \left(\frac{5\pi}{6} + 1\right)\hat{j}$$

$$\vec{v}_{BA} = -5\pi\hat{i} - \hat{j}$$

$$v_{_{BA}}=\sqrt{25\pi^2+1}$$

(III)
$$x_A = \sin t$$

$$v_A = \cos t = \frac{1}{2} m / s$$

$$x_{B} = \cos t$$

$$v_B = -\sin t = -\frac{\sqrt{3}}{2} m/s$$

$$v_{BA} = -\frac{\sqrt{3}}{2} - \frac{1}{2}$$

(IV) $\vec{v}_A \& \vec{v}_B$ are always perpendicular

So,
$$|\vec{V}_{BA}| = \sqrt{V_A^2 + V_B^2} = \sqrt{10} \text{m/s}$$

17. List I describes thermodynamic processes in four different systems. List II gives the magnitudes (either exactly or as a close approximation) of possible changes in the internal energy of the system due to the process.

List I List II

- (I) 10⁻³ kg of water at 100°C is converted to steam at the same

 (P) 2 kJ

 Temperature, at a pressure of 10⁵ Pa. The volume of the system

 From 10⁻⁶ m³ to 10⁻³ m³ in the process. Latent heat of water

 = 2250 kJ/kg.
- (II) 0.2 moles of a rigid diatomic ideal gas with volume V at temperature (Q) 7 kJ 500 K undergoes an isobaric expansion to volume 3 V. Assume $R = 8.0 \text{ J mol}^{-1} \text{ K}^{-1}.$
- (III) On mole of a monatomic ideal gas is compressed adiabatically from $(R) \ 4 \ kJ$ volume $\ V = \frac{1}{3} m^3 \$ and pressure 2 kPa to volume $\frac{\nu}{8}$
- (IV) Thee moles of a diatomic ideal gas whose molecules can vibrate, (R) 5 kJ is given 9 kJ of heat and undergoes isobaric expansion.

Which one of the following options is correct?

(T) 3 kJ

(A) I - T, II - R, III - S, IV - Q

(B) I - S, II - P, III - T, IV - P

(C) I - P, II - R, III - T, IV - Q

(D) I - Q, II - R, III - S, IV - T

Ans. (C)

Sol. (I)
$$\Delta U = \Delta Q - \Delta W$$

$$= \left\{ \left(10^{-3} \times 2250\right) - \frac{10^{5} \left(10^{-3} - 10^{-6}\right)}{10^{3}} \right\} kJ$$

$$= (2.25 - 0.0999) \text{ kJ}$$

$$= (2.1501) kJ$$

(II)
$$\Delta U = nC_{\nu}\Delta T$$

$$=\frac{5}{2}nR\Delta T$$

$$=\frac{5}{2}\cdot(0.2)(8)(1500-500)J$$

$$= 4 kJ$$

(III)
$$P_1V_2^{\gamma} = P_2V_2^{\gamma}$$

$$\Rightarrow 2\left(\frac{1}{3}\right)^{5/3} = P_2\left(\frac{1}{24}\right)^{5/3}$$

$$\Rightarrow$$
 P₂ = 64 kPa

$$\Delta U = nC_{V}\Delta T = \frac{3}{2} \cdot \left(P_{2}V_{2} - P_{1}V_{1}\right)$$

$$=\frac{3}{2}\bigg(64\times\frac{1}{24}-2\times\frac{1}{3}\bigg)kJ$$

$$= 3 kJ$$

(IV)
$$\Delta U = nC_v \Delta T$$

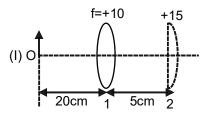
$$= n \cdot \frac{7}{2} R \Delta T$$

$$=\frac{7}{9}\Delta Q$$

$$= 7 \text{ kJ}$$

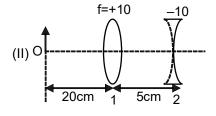
18. List I contains four combinations of two lenses (1 and 2) whose focal lengths (in cm) are indicated in the figures. In all cases, the object is placed 20 cm from the first lens on the left, and the distance between the two lenses is 5 cm. List II contains the positions of the final images.

List I

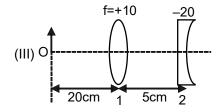


List II

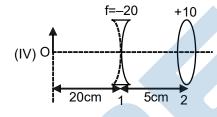
(P) Final image is farmed at 7.5 cm on the right side of lens 2.



(Q) Final image is farmed at 60.0 cm on the right side of lens 2.



(R) Final image is farmed at 30.0 cm on the left side of lens 2.



(S) Final image is farmed at 6.0 cm on the right side of lens 2.

(T) Final image is farmed at 30.0 cm on the right side of lens 2.

Which one of the following options is correct?

Ans. (A)

$$\textbf{Sol.} \quad \textbf{(I)} \ \ v_1 = \frac{uf}{u+f}$$

$$=\frac{(-20)(10)}{(-20)+(10)}=+20$$

$$u_2 = +15$$

$$v_2 = \frac{(15)(15)}{(15) + (15)} = +7.5$$

$$(II) v_1 = +20$$

$$u_2 = +15$$

$$v_2 = \frac{(15)(-10)}{(15) + (-10)} = -30$$

$$(III) v_1 = +20$$

$$u_2 = +15$$

$$v_2 = \frac{(15)(-20)}{(15) + (-20)} = 60$$

(IV)
$$v_1 = \frac{(-20)(-20)}{(-20) + (-20)} = -10$$

$$u_2 = -15$$

$$v_2 = \frac{(-15)(10)}{(-15) + (10)} = 30$$



PART B: CHEMISTRY

SECTION-1: (Maximum Marks: 24)

- This section contains Eight (08) question stems.
- The answer to each question is a **NUMERICAL VALUE**.
- For each question, enter the correct numerical value of the answer using the mouse and the
- on-screen virtual numeric keypad in the place designated to enter the answer. If the numerical
- value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 **ONLY** if the correct numerical value is entered at the designated place;

Zero Marks : 0 In all other cases.

2 mol of Hg(g) is combusted in a fixed volume bomb calorimeter with excess of O₂ at 298 K and 1 atm into HgO(s). During the reaction, temperature increases from 298.0 K to 312.8 K. If heat capacity of the bomb calorimeter and enthalpy of formation of Hg(g) are 20.00 kJ K⁻¹ and 61.32 kJ mol⁻¹ at 298 K, respectively, the calculated standard molar enthalpy of formation of HgO(s) at 298 K is X kJ mol⁻¹. The value of |X| is ______.

[Given : Gas constant $R = 8.3 \text{ J K}^{-1} \text{ mol}^{-1}$]

Ans. (90.39)

Sol.
$$Q_{rxn} = C\Delta T$$

$$|\Delta U| \times 2 = 20 \times 14.8$$

$$\Delta U = 148 \text{ kJ/mol}$$

$$Hg(g) + \frac{1}{2}O_2(g) \longrightarrow HgO(s) : \Delta U = -148kJ / mol$$

$$\Delta H = \Delta U + Dn_{g}RT$$

$$=-148-\frac{3}{2}\times\frac{8.3}{1000}\times298=-151.7101$$

$$Hg(I) + \frac{1}{2}O_2(g) \longrightarrow HgO(s)$$

$$\Delta H = -151.7101 + 61.32 = -90.39 \text{ kJ/mol}$$

2. The reduction potential (E^0 , in V) of $MnO_4^-(aq)/Mn(s)$ is _____.

$$[Given: \ E^0_{\left(MnO_4^-(aq)/MnO_2(s)\right)} = 1.68V; E^0_{\left(MnO_2(s)/Mn^{2^+}(aq)\right)} = 1.21V; E^0_{\left(Mn^{2^+}(aq)/Mn(s)\right)} = -1.03V \]$$

Ans. (0.77)

Sol.
$$MnO_4^{-\frac{(3)}{4}} \xrightarrow{\text{M}} O_2 \xrightarrow{\text{(2)}} Mn^{+2} \xrightarrow{\text{(2)}} Mn$$

For the required reaction $\Delta G^{\circ} = \Delta G^{\circ}_{1} + \Delta G^{\circ}_{2} + \Delta G^{\circ}_{3}$

$$\Rightarrow$$
 7 × E = 1.68 × 3 + 1.21 × 2 + (-1.03) × 2

$$E = \frac{5.4}{7} = 0.7714$$

3. A solution is prepared by mixing 0.01 mol each of H₂CO₃, NaHCO₃, Na₂CO₃, and NaOH in 100 mL of water. pH of the resulting solution is _____.

[Given : pK_{a1} and pK_{a2} of H_2CO_3 are 6.37 and 10.32, respectively ; log 2 = 0.30]

Ans. (10.02)

Sol.
$$H_2CO_3 + NaOH \longrightarrow NaHCO_3 + H_2O$$

Milli moles 10 10 –

At end 0 0 10 + 10 = 20

Final mixture has 20 milli moles NaHCO₃ and 10 milli moles Na₂CO₃

$$pH = pKa_2 + log \frac{Salt}{Acid}$$

$$pH = pKa_2 + log \frac{Salt}{Acid}$$
 [Buffer : $Na_2CO_3 + NaHCO_3$]

$$= 10.32 - \log 2 = 10.02$$

4. The treatment of an aqueous solution of 3.74 g of $Cu(NO_3)_2$ with excess KI results in a brown solution along with the formation of a precipitate. Passing H_2S through this brown solution gives another precipitate X. The amount of X (in g) is

[Given : Atomic mass of H = 1, N = 14, O = 16, S = 32, K = 39, Cu = 63, I = 127]

Ans. (0.32)

Sol.
$$2Cu(NO_3)_2 + 5KI \longrightarrow Cu_2I_2 + KI_3 + 4KNO_3$$

0.02

$$KI_3 + H_2S \longrightarrow S \downarrow + KI + 2HI$$

0.01 0.01

 $n_s = 0.01 \text{ mole}$

weight of sulphur = $32 \times 0.01 = 0.32$ gm

5. Dissolving 1.24 g of white phosphorous in boiling NaOH solution in an inert atmosphere gives a gas Q.

The amount of $CuSO_4$ (in g) required to completely consume the gas Q is _____.

[Given : Atomic mass of H = 1, O = 16, Na = 23, P = 31, S = 32, Cu = 63]

Ans. (2.38 - 2.39)

Sol. Mole of
$$P_4 = \frac{1.24}{31 \times 4} = 0.01$$

$$P_4 + 3NaOH + 3H_2O \longrightarrow PH_3 + 3NaH_2PO_2$$

0.01 mole 0.01 mole

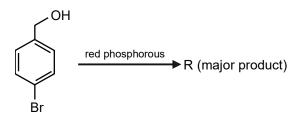
$$2PH_3 + 3CuSO_4 \rightarrow Cu_3P_2 + 3H_2SO_4$$

$$0.01 \frac{3}{2} \times 0.01$$

$$=\frac{0.03}{2}$$
 moles

$$W_{\text{CuSO}_4} = \frac{0.03}{2} \times 159 = 2.385 gm$$

6. Consider the following reaction.



On estimation of bromine in 1.00 g of R using Carius method, the amount of AgBr formed (in g) is

[Given : Atomic mass of H = 1, C = 12, O = 16, P = 31, Br = 80, Ag = 108]

Ans. (1.50)

Sol. Red P Br Br M.W. = 250 g/mol

$$1gR \rightarrow \frac{1}{250}$$
 moles

No. of Br atoms $\rightarrow \frac{2}{250}$ moles

Mole of AgBr $\rightarrow \frac{2}{250}$ moles

Mass of AgBr =
$$\frac{2}{250} \times (108 + 80) = 1.504$$

7. The weight percentage of hydrogen in Q, formed in the following reaction sequence, is _____.

[Given : Atomic mass of H = 1, C = 12, N = 14, O = 16, S = 32, Cl = 35]

Ans. (1.31)

Sol.
$$\begin{array}{c|c} CI & ONa & OH \\ \hline NaOH, 623 \, K & O_2N & O_2N \\ \hline NaOH, 623 \, K & O_2N & O_2N \\ \hline And & O_2N & O_2N \\ \hline And & O_2N & O_2N \\ \hline O_2N & O_2N & O_2N$$

8. If the reaction sequence given below is carried out with 15 moles of acetylene, the amount of the product D formed (in g) is _____.

$$HC \equiv CH \xrightarrow{\text{iron tube (red hot)}} A \xrightarrow{H_3C} CI \qquad B \xrightarrow{1. O_2} CH_3COCI \qquad D$$

$$(80\%) \qquad (50\%) \qquad CCH_3COCH_3 \qquad C$$

$$(50\%) \qquad (100\%)$$

The yields of A, B, C and D are given in parentheses.

[Given : Atomic mass of H = 1, C = 12, O = 16, Cl = 35]

Ans. (136)

Sol. HC = CH
$$\frac{Iron}{red hot}$$
 $\frac{H_3C}{AICI_3}$ $\frac{1. O_2}{2. H_3O^{\circ}}$ $\frac{1. mol}{50\% \text{ yield}}$ $\frac{1. mol}{50\% \text{ yield}}$ $\frac{1. mol}{50\% \text{ yield}}$ $\frac{1. mol}{100\% \text{ yield}}$ $\frac{1. mol}{10$

SECTION-2: (Maximum Marks: 24)

- This section contains SIX (06) questions.
- Each question has **FOUR** options (A), (B), (C) and (D). **ONE OR MORE THAN ONE** of these four option(s) is (are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** (all) the correct option(s) is(are) chosen;

• Partial Marks : +3 If all the four options are correct but **ONLY** three options are chosen;

• Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen,

both of which are correct;

Partial Marks : +1 If two or more options are correct but ONLY one option is chosen and it

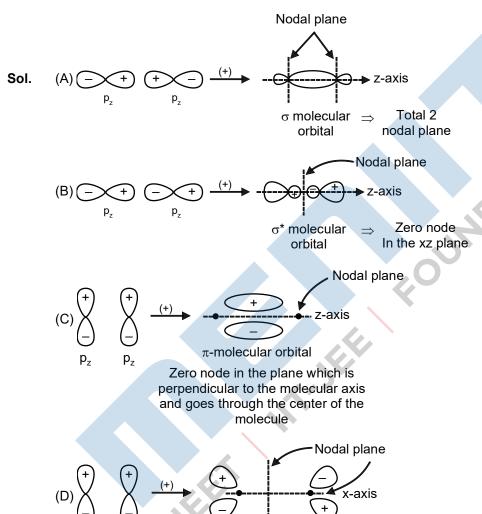
is a correct option;

Zero Marks : 0 If unanswered;

Negative Marks : -2 In all other cases.

- **9.** For diatomic molecules, the correct statement(s) about the molecular orbitals formed by the overlap to two 2p_z orbitals is(are)
 - (A) σ orbital has a total of two nodal planes.
 - (B) σ^* orbital has one node in the xz-plane containing the molecular axis.
 - (C) π orbital has one node in the plane which is perpendicular to the molecular axis and goes through the center of the molecule.
 - (D) π^* orbital has one node in the xy-plane containing the molecular axis.

Ans. (A,D)



the molecular axis

10. The correct option(s) related to adsorption processes is(are)

 p_z

- (A) Chemisorption results in a unimolecular layer.
- (B) The enthalpy change during physisorption is in the range of 100 to 140 kJ mol⁻¹.

π*-moleċular orbital

One node in xy plane containing

- (C) Chemisorption is an endothermic process.
- (D) Lowering the temperature favors physisorption processes.

Ans. (A,D)

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- **Sol.** (A) Chemisorption is unimolecular layered.
 - (B) Enthalpy of physisorption is much less in magnitude.
 - (C) Chemisorption of gases on solids is exothermic.
 - (D) As physisorption is exothermic so lowering temperature favours it.
- **11.** The electrochemical extraction of aluminum from bauxite ore involves.
 - (A) the reaction of Al_2O_3 with coke (C) at a temperature > 2500°C.
 - (B) the neutralization of aluminate solution by passing CO_2 gas to precipitate hydrated alumina $(Al_2O_3.3H_2O)$
 - (C) the dissolution of Al₂O₃ in hot aqueous NaOH.
 - (D) the electrolysis of Al₂O₃ mixed with Na₃AlF₆ to give Al and CO₂.

Ans. (B,C,D)

Sol. (A) Electrochemical extraction of Aluminum from bauxite done below 2500°C

(B) $2Na[Al(OH)_4]aq. + 2CO_2(g) \rightarrow Al_2O_3.3H_2O(s) \downarrow + 2NaHCO_3(aq.)$

The sodium aluminate present in solution is neutralised by passing CO₂ gas and hydrated Al₂O₃ is precipitated.

(C) $Al_2O_3(s) + 2NaOH(aq.) + 3H_2O(I) \rightarrow 2Na[Al(OH)_4]aq.$

Concentration of bauxite is carried out by heating the powdered ore with hot concentrated solution of NaOH

- (D) In metallurgy of aluminum, Al₂O₃ is mixed with Na₃AlF₆
- 12. The treatment of galena with HNO₃ produces a gas that is
 - - (B) bent in geometry (C) an acidic oxide (D) colorless

Ans. (A,D)

(A) paramagnetic

Sol. $3PbS + 8HNO_3 \rightarrow 3Pb(NO_3)_2 + 2NO + 4H_2O + S$

NO

Neutral oxide, Paramagnetic, Linear geometry, Colourless gas

13. Considering the reaction sequence given below, the correct statement(s) is(are)

$$H_3C$$
 COOH $\frac{1. Br_2, red phosphorous}{2. H_2O}$ P $\frac{1. \bigvee_{Q} N^{\Theta}K^{\Theta}}{2. NaOH}$ Q + COOH

- (A) P can be reduced to a primary alcohol using NaBH₄.
- (B) Treating P with conc. NH₄OH solution followed by acidification gives Q.
- (C) Treating Q with a solution of NaNO₂ in aq. HCl liberates N₂.
- (D) P is more acidic than CH₃CH₂COOH

Ans. (B,C,D)

Sol.
$$H_3C$$
 $COOH$ $\frac{1. \operatorname{Red} P, \operatorname{Br}_2}{2. H_2O}$ $COOH$ $COOH$ OOH OO

. . .

COOH+N₂

14. Consider the following reaction sequence,

$$H_3C$$
 P
 Q
 R
 H_2O
 T
 U
 $COOH$

the correct option(s) is(are)

(A)
$$P = H_2/Pd$$
, ethanol

2. KMnO₄ – KOH, heat

$$S = \bigcup_{\mathsf{H}_3\mathsf{C}} \mathsf{N}_2\mathsf{Cl}^\Theta$$

(C)
$$S = \bigcup_{H_3C} N_2CI^{\Theta}$$

$$U = 1. CH3CH2OH$$

2. KMnO₄ – KOH, heat

(D) Q =
$$R = H_2/Pd$$
, ethanol

$$T = \bigcup_{H_3C} OH$$

Ans. (A,B,C)

SECTION-3: (Maximum Marks: 12)

- This section contains FOUR (04) Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has TWO lists: List-I and List-II.
- List-I has Four entries (I), (II), (III) and (IV) and List-II has Five entries (P), (Q), (R), (S) and (T).
- FOUR options are given in each Multiple Choice Question based on List-I and List-II and ONLY ONE
 of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated <u>according to the following marking scheme</u>:

Full Marks : +3 ONLY if the option corresponding to the correct combination is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

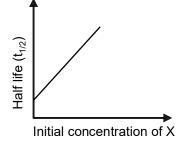
Negative Marks : -1 In all other cases.

15. Match the rate expressions in LIST-I for the decomposition of X with the corresponding profiles provided in LIST-II. X_s and k constants having appropriate units.

(P)

List II

(I) rate =
$$\frac{k[X]}{X_s + [X]}$$



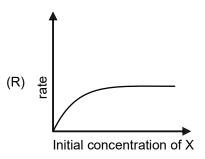
Under all possible initial concentration of X

(II) rate =
$$\frac{k[X]}{X_s + [X]}$$
 (Q) $\frac{\int_{\frac{1}{2}}^{\frac{1}{2}}}{\int_{\frac{1}{2}}^{\frac{1}{2}}}$ Initial concentration of X

Where initial concentration of X are much

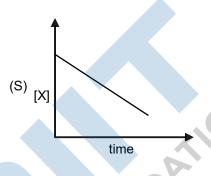
less than X_s

(III) rate =
$$\frac{k[X]}{X_s + [X]}$$

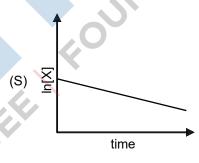


Where initial concentration of X are much higher than $\boldsymbol{X}_{\!s}$

(IV) rate =
$$\frac{k[X]^2}{X_s + [X]}$$



Where initial concentration of X is much higher than $\boldsymbol{X}_{\!s}$



Ans. (A

Sol. (I) rate=

$$rate = \frac{k[x]}{x_s + [x]} = \frac{k}{\frac{x_s}{[x]} + 1}$$

If $[X] \rightarrow \infty \Rightarrow \text{rate} \rightarrow k \Rightarrow \text{order} = 0$

$$\Rightarrow$$
 (I) – (R), (P)

(II) $[x] \ll x_s \Rightarrow rate = \frac{k[x]}{x_s} \Rightarrow order = 1$

$$\Rightarrow$$
 (II) $-$ (Q), (T)

(III) $[x] > x_s \Rightarrow \text{rate} = k \Rightarrow \text{order} = 0$

$$\Rightarrow$$
 (III) – (P), (S)

(IV) rate =
$$\frac{k[x]^2}{x_s + [x]}$$

$$[x] > x_s \Rightarrow \text{rate} = k[x]$$

$$\Rightarrow$$
 (IV) – (Q), (T)

16. LIST-I contains compounds and LIST-II contains reaction

LIST-I

LIST-II

(I) H_2O_2

(P) Mg(HCO₃)₂ + Ca(OH)₂ \rightarrow

(II) $Mg(OH)_2$

(Q) BaO₂ + $H_2SO_4 \rightarrow$

(III) BaCl₂

 $(R) Ca(OH)_2 + MgCl_2$

(IV) CaCO₃

(S) $BaO_2 + HCI \rightarrow$

(T) Ca(HCO₃)₂ + Ca(OH)₂ \rightarrow

Match each compound in LIST – I with its formation reaction(s) in LIST-II, and choose the correct option

Ans. (D)

Sol. (P) $Mg(HCO_3)_2 + 2Ca(OH)_2 \rightarrow Mg(OH)_2 + 2CaCO_3 + 2H_2O$

(Q) $BaO_2 + H_2SO_4 \rightarrow H_2O_2 + BaSO_4$

(R) $Ca(OH)_2 + MgCl_2 \rightarrow Mg(OH)_2 + CaCl_2$

(S) $BaO_2 + 2HCI \rightarrow BaCl_2 + H_2O_2$

(T) Ca(HCO₃)₂ + Ca(OH)₂ \rightarrow 2CaCO₃ + 2H₂O

17. LIST-I contains metal species and LIST-II contains their properties.

LIST-I

LIST-II

(I) $[Cr(CN)_6]^{4-}$

(P) t_{2q} orbitals contain 4 electrons

(II) [RuCl₆]²⁻

(Q) μ (spin-only) = 4.9 BM

(III) [Cr(H \bigcirc) 1^{2+}

(R) low spin complex ion

(IV) [Ea/H O) 1²⁴

(S) metal ion in 4+ oxidation state

(T) d₄ species

[Given: Atomic number of Cr = 24, Ru = 44, Fe = 26]

Metal each metal species in LIST-I with their properties in LIST-II, and choose the correct option

(A) I - R, T; II - P, S; III - Q, T; IV - P, Q

(B) I - R, S; II - P, T; III - P, Q; IV - Q, T

(C) I - P, R; II - R, S; III - R, T; IV - P, T

(D) I - Q, T; II - S, T; III - P, T; IV - Q, R

Ans. (A)

Sol. (1) $[Cr(CN)_6]^{4-}$

 $Cr^{+2} = [Ar]_{18} 3d^4 4s^0$; low spin complex

- P, R, T
- (2) $[RuCl_6]^{2-}$

 $Ru^{+4} = [Kr]_{36}4d^45s^0$; low spin complex



- P, R, S, T
- (3) $[Cr(H_2O)_6]^{2+}$

 $Cr^{+2} = [Ar]_{18}3d^44s^0$; high spin complex



- Q, T
- $(4) [Fe(H_2O)_6]^{2+}$

 $Fe^{+2} = [Ar]_{18}3d^6$; High spin complex

$$\frac{1}{2} \quad \frac{1}{4} \quad \Delta_0
$$\frac{1}{2} \quad \frac{1}{4} \quad \frac{1}{4} \quad t_2^4$$$$

- P. C
- 18. Match the compounds in LIST-I with the observation in LIST-II, and choose the correct option.

LIST-I

LIST-II

- (I) Aniline
- (P) Sodium fusion extract of the compound on boiling with FeSO₄, followed by acidification with conc. H₂SO₄, gives Prussian blue color.
- (II) o-Cresol
- (Q) Sodium fusion extract of the compound on treatment with sodium
- nitroprusside gives blood red color.
- (III) Cysteine
- (R) Addition of the compound to a saturated solution of NaHCO_3 results in
- effervescence.
- (IV) Coprolactam
- (S) The compound reacts with bromine water to give a white precipitate.
- (T) Treating the compound with neutral FeCl₃ solution produces violet color.
- (A) I P, Q; II S; III Q, R; IV P
- (B) I P; II R, S; III R; IV Q, S
- (C) I Q, S; II P, T; III P; IV S
- (D) I P, S; II T; III Q, R; IV P

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Ans. (D)

Sol. Sol. : Blue colour in Lassign test due to presence of N
Aniline

Me
OH
: Violet colour with FeCl₃ due to presence of phenolic OH
o-Cresol

HS-CH₂-CH-COOH : It gives blod red colour with NaSCN NH₂ Cystein

N–H
O : Blue colour in Lassign test due to presence of N

Caprolactam

PART C: MATHEMATICS

SECTION-1: (Maximum Marks: 24)

- This section contains **Eight (08)** question stems.
- The answer to each question is a **NUMERICAL VALUE**.
- For each question, enter the correct numerical value of the answer using the mouse and the
- on-screen virtual numeric keypad in the place designated to enter the answer. If the numerical
- value has more than two decimal places, truncate/round-off the value to TWO decimal places.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 **ONLY** if the correct numerical value is entered at the designated place;

A.JEE

FOUNDATIO!

Zero Marks : 0 In all other cases.

1. Considering only the principal values of the inverse trigonometric functions, the value of

$$\frac{3}{2}\cos^{-1}\sqrt{\frac{2}{2+\pi^2}} + \frac{1}{4}\sin^{-1}\frac{2\sqrt{2}\pi}{2+\pi^2} + \tan^{-1}\frac{\sqrt{2}\pi}{\pi}$$

is _____

Ans. (2.35 or 2.36)

Sol.
$$\cos^{-1} \sqrt{\frac{2}{2+\pi^2}} = \tan^{-1} \frac{\pi}{\sqrt{2}}$$



$$\sin^{-1}\left(\frac{2\sqrt{2}\pi}{2+\pi^2}\right) = \sin^{-1}\left(\frac{2\times\frac{\pi}{\sqrt{2}}}{1+\left(\frac{\pi}{\sqrt{2}}\right)^2}\right)$$

$$= \pi - 2 \tan^{-1} \left(\frac{\pi}{\sqrt{2}} \right)$$

$$\left(\text{ As, } \sin^{-1} \left(\frac{2x}{1+x^2} \right) = \pi - 2 \tan^{-1} x, x \ge 1 \right)$$

and
$$\tan^{-1} \frac{\sqrt{2}}{\pi} = \cot^{-1} \left(\frac{\pi}{\sqrt{2}} \right)$$

$$\therefore \text{Expression} = \frac{3}{2} \left(tan^{-1} \frac{\pi}{\sqrt{2}} \right) + \frac{1}{4} \left(\pi - 2 tan^{-1} \frac{\pi}{\sqrt{2}} \right) + \cot^{-1} \left(\frac{\pi}{\sqrt{2}} \right)$$

$$= \left(\frac{3}{2} - \frac{2}{4}\right) tan^{-1} \frac{\pi}{\sqrt{2}} + \frac{\pi}{4} + \cot^{-1} \frac{\pi}{\sqrt{2}}$$

$$= \left(tan^{-1} \frac{\pi}{\sqrt{2}} + cot^{-1} \frac{\pi}{\sqrt{2}} \right) + \frac{\pi}{4}$$

$$=\frac{\pi}{2}+\frac{\pi}{4}=\frac{3\pi}{4}$$

2. Let α be a positive real number. Let $f: \mathbb{R} \to \mathbb{R}$ and $g: (\alpha, \infty) \to \mathbb{R}$ be the functions defined by

$$f(x) = sin\left(\frac{\pi x}{12}\right) and g(x) = \frac{2 \log_{e}(\sqrt{x} - \sqrt{\alpha})}{\log_{e}\left(e^{\sqrt{x}} - e^{\sqrt{\alpha}}\right)}$$

Then the value of $\lim_{x\to a^+} f(g(x))$ is _____

Ans. (0.50)

$$\text{Sol.} \qquad \lim_{x \to a^+} \frac{2 \ln(\sqrt{x} - \sqrt{\alpha})}{\ln\left(e^{\sqrt{x}} - e^{\sqrt{\alpha}}\right)} \left(\frac{0}{0} \text{form}\right)$$

.. Using Lopital rule,

$$=2\lim_{x\to a^{+}}\frac{\left(\frac{1}{\sqrt{x}-\sqrt{\alpha}}\right)\cdot\frac{1}{2\sqrt{x}}}{\left(\frac{1}{e^{\sqrt{x}}-e^{\sqrt{a}}}\right)\cdot e^{\sqrt{x}}\cdot\frac{1}{2\sqrt{x}}}$$

$$= \frac{2}{e^{\sqrt{\alpha}}} \lim_{x \to a^+} \!\! \frac{\left(e^{\sqrt{x}} - e^{\sqrt{\alpha}}\right)}{\left(\sqrt{x} - \sqrt{\alpha}\right)} \!\! \left(\frac{0}{0}\right)$$

$$=\frac{2}{e^{\sqrt{x}}} \lim_{x \to a^+} \frac{\left(e^{\sqrt{x}} \cdot \frac{1}{2\sqrt{x}} - 0\right)}{\left(\frac{1}{2\sqrt{x}} - 0\right)} = 2$$

So,
$$\lim_{x\to a^+} f(g(x)) = \lim_{x\to a^+} f(2)$$

$$=f(2)=sin\frac{\pi}{6}=\frac{1}{2}$$

$$= 0.50$$

- 3. In a study about a pandemic, data of 900 persons was collected. It was found that
 - 190 persons had symptom of fever,
 - 220 persons had symptom of cough,
 - 220 persons had symptom of breathing problem,
 - 330 persons had symptom of fever or cough or both,
 - 350 persons had symptom of cough or breathing problem or both,
 - 340 persons had symptom of fever or breathing problem or both,
 - 30 persons had all three symptoms (fever, cough and breathing problem).

If a person is chosen randomly from these 900 persons, then the probability that the person has at most one symptom is _____.

Ans. (0.80)

Sol.
$$n(U) = 900$$

Let $A \equiv Fever$, $B \equiv Cough$

C ≡ Breathing problem

$$\therefore$$
 n(A) = 190, n(B) = 220, n(C) = 220

$$n(A \cup B) = 330, n(B \cup C) = 350,$$

$$n(A \cup C) = 340, n (A \cap B \cap C) = 30$$

Now
$$n(A \cup B) = n(A) + n(B) - n(A \cap B)$$

$$\Rightarrow$$
 330 = 190 + 220 - n(A \cap B)

$$\Rightarrow$$
 n(A \cap B) = 80

Similarly,

$$350 = 220 + 220 - n(B \cap C)$$

$$\Rightarrow$$
 n(B \cap C) = 90

and
$$340 = 190 + 220 - n(A \cap C)$$

$$\Rightarrow$$
 n(A \cap C) = 70

$$\therefore$$
 n(A \cup B \cup C) = (190 + 220 + 220) – (80 + 90 + 70) + 30

$$= 660 - 240 = 420$$

⇒ Number of person without any symptom

$$= n (\cup) - n(A \cup B \cup C)$$

$$= 900 - 420 = 480$$

Now, number of person suffering from exactly one symptom

$$= (n(A) + n(B) + n(C)) - 2(n(A \cap B) + n(B \cap C) + n(C \cap A)) + 3n(A \cap B \cap C)$$

$$= (190 + 220 + 220) - 2(80 + 90 + 70) + 3(30)$$

$$= 630 - 480 + 90 = 240$$

.. Number of person suffering from atmost one symotom

$$\Rightarrow$$
 Probability = $\frac{720}{900} = \frac{8}{10} = \frac{4}{5} = 0.80$

4. Let z be a complex number with non-zero imaginary part. If $\frac{2+3z+4z^2}{2-3z+4z^2}$ is a real number, then the value of $|z|^2$ is

Ans. (0.50)

Sol. Given that

 $Z \neq \overline{Z}$

Let
$$\alpha = \frac{2+3z+4z^2}{2-3z+4z^2} = \frac{\left(2-3z+4z^2\right)+6z}{2-3z+4z^2}$$

$$\therefore \alpha = 1 + \frac{6z}{2 - 3z + 4z^2}$$

If α is a real number, then

$$\alpha = \overline{\alpha}$$

$$\Rightarrow \frac{z}{2-3z+4z^2} = \frac{\overline{z}}{2-3\overline{z}+4\overline{z}^2}$$

$$\therefore 2(z-\overline{z})=4z\overline{z}(z-\overline{z})$$

As
$$z \neq \overline{z}$$
 (Given)

$$\Rightarrow z\overline{z} = \frac{2}{4} = \frac{1}{2}$$

$$\Rightarrow |z|^2 = 0.50$$

- 5. Let \overline{z} denote the complex conjugate of a complex number z and let $i = \sqrt{-1}$. In the set of complex numbers, the number of distinct roots of the equation $\overline{z} z^2 = i(\overline{z} + z^2)$ is _____.
- **Ans.** (4.00)
- Sol. Given,

$$\overline{z} - z^2 = i(\overline{z} + z^2)$$

$$\Rightarrow (1-i)\overline{z} = (1+i)z^2$$

$$\Rightarrow \frac{(1-i)}{(1+i)}\overline{Z} = Z^2$$

$$\Rightarrow \left(-\frac{2i}{2}\right)\overline{z} = z^2$$

$$\therefore Z^2 = -i\overline{Z}$$

Let
$$z = x + iy$$
,

:
$$(x^2 - y^2) + i(2xy) = -i(x - iy)$$

so,
$$x^2 - y^2 + y = 0$$
 ...(1)

and
$$(2y + 1)x = 0$$
 ...(2)

$$\Rightarrow$$
 x = 0 or y = $-\frac{1}{2}$

Case I: When x = 0

$$\therefore$$
 (1) \Rightarrow y(1 - y) = 0 \Rightarrow y = 0,1

Case II: When $y = -\frac{1}{2}$

$$\therefore (1) \Rightarrow x^2 - \frac{1}{4} - \frac{1}{2} = 0 \Rightarrow x^2 = \frac{3}{4} \Rightarrow x = \pm \frac{\sqrt{3}}{2}$$

$$\therefore \left(\frac{\sqrt{3}}{2}, -\frac{1}{2}\right), \left(-\frac{\sqrt{3}}{2}, -\frac{1}{2}\right)$$

- ⇒ Number of distinct 'z' is equal to 4.
- 6. Let ℓ_1 , ℓ_2 ,, ℓ_{100} be consecutive terms of an arithmetic progression with common difference d_1 , and let ℓ_1 , ℓ_2 ,, ℓ_{100} be consecutive terms of another arithmetic progression with common difference d_2 , where d_1d_2 = 10. For each i = 1, 2,···,100, let R_i be a rectangle with length ℓ_i , width w_i and area A_i . If A_{51} A_{50} = 1000, then the value of A_{100} A_{90} is ______.

Ans. (18900.00)

Sol. Given

$$A_{51} - A_{50} = 1000 \Rightarrow \ell_{51} W_{51} - \ell_{50} W_{50} = 1000$$

$$\Rightarrow$$
 $(\ell_1 + 50d_1) (w_1 + 50d_2) - (\ell_1 + 49d_1) (w_1 + 49d_2) = 1000$

$$\Rightarrow$$
 ($\ell_1 d_2 + w_1 d_1$) = 10

(As
$$d_1d_2 = 10$$
)

$$\therefore A_{100} - A_{90} = \ell_{100} W_{100} - \ell_{90} W_{90}$$

$$= (\ell_1 + 99d_1) (w_1 + 99d_2) - (\ell_1 + 89d_1) (w_1 + 89d_2)$$

= 10
$$(\ell_1 d_2 + w_1 d_1) + (99^2 - 89^2) d_1 d_2$$

$$=10(10)+(99-89)(99+89)(10)$$

(As
$$d_1d_2 = 10$$
)

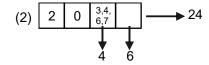
$$= 100 (1 + 188) = 100 (189)$$

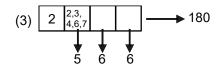
7. The number of 4-digit integers in the closed interval [2022, 4482] formed by using the digits 0, 2, 3, 4, 6, 7 is ______.

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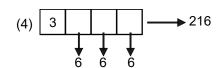
Ans. (569.00)

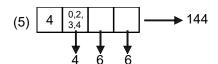
Sol. (1)
$$\begin{bmatrix} 2 & 0 & 2 & 2,3, \\ 4,6,7 & & & & \end{bmatrix}$$





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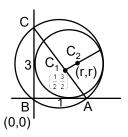


Number of 4 digit integers in [2022,4482]

Let ABC be the triangle with AB = 1, AC = 3 and $\angle BAC = \frac{\pi}{2}$. If a circle of radius r > 0 touches the sides 8. AB, AC and also touches internally the circumcircle of the triangle ABC, then the value of r is

(0.83 or 0.84) Ans.

 $4 - \sqrt{10} = 0.83$ or 0.84Sol.



$$C_1\left(\frac{1}{2}, \frac{3}{2}\right)$$
 and $r_1 = \frac{\sqrt{10}}{2}$

$$C_2 = (r,r)$$

∴ circle C₂ touches C₁ internally

$$\Rightarrow C_1 C_2 = \left| r - \frac{\sqrt{10}}{2} \right|$$

$$(0.83 \text{ or } 0.84)$$

$$4 - \sqrt{10} = 0.83 \text{ or } 0.84$$

$$C$$

$$3$$

$$C_1 = \sqrt{\frac{1}{2}} \sqrt{\frac{3}{2}} \sqrt{r,r}$$

$$C_1 = \sqrt{\frac{1}{2}} \sqrt{\frac{3}{2}} \sqrt{r}$$

$$C_2 = (r,r)$$

$$\therefore \text{ circle } C_2 \text{ touches } C_1 \text{ internally}$$

$$\Rightarrow C_1 C_2 = \left| r - \frac{\sqrt{10}}{2} \right|$$

$$\Rightarrow \left(r - \frac{1}{2} \right)^2 + \left(r - \frac{3}{2} \right)^2 = \left(r - \frac{\sqrt{10}}{2} \right)^2$$

$$r^2 - 4r + \sqrt{10}r = 0$$

r = 0 (rejected) or $r = 4 - \sqrt{10}$

SECTION-2: (Maximum Marks: 24)

- This section contains SIX (06) questions.
- Each question has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is (are) correct answer(s).
- For each question, choose the option(s) corresponding to (all) the correct answer(s).
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +4 If **ONLY** (all) the correct option(s) is(are) chosen;

- : +3 If all the four options are correct but **ONLY** three options are chosen; Partial Marks
- Partial Marks : +2 If three or more options are correct but **ONLY** two options are chosen,

both of which are correct;

: +1 If two or more options are correct but ONLY one option is chosen and it **Partial Marks**

is a correct option;

- Zero Marks : 0 If unanswered;
- **Negative Marks** : -2 In all other cases.
- $\text{Consider the equation} \int\limits_{1}^{e} \frac{\left(\log_{e}x\right)^{1/2}}{x\left(a-\left(\log_{e}x\right)^{3/2}\right)^{2}} dx = 1, \quad a \in (-\infty,0) \cup (1,\infty)\,.$ **FOUNDA** 9.

Which of the following statements is/are TRUE?

- (A) No a satisfies the above equation
- (B) An integer a satisfies the above equation
- (C) An irrational number a satisfies the above equation
- (D) More than one a satisfy the above equation

Ans. (C,D)

Sol.
$$\int_{1}^{e} \frac{\left(\log_{e} x\right)^{1/2}}{x\left(a - \left(\log_{e} x\right)^{3/2}\right)^{2}}$$

Let
$$a - (\log_e x)^{3/2} = t$$

$$\frac{\left(\log_{e} x\right)^{1/2}}{x} dx = -\frac{2}{3} dt$$

$$= \frac{2}{3} \int_{a}^{a-1} \frac{-dt}{t^2} = \frac{2}{3} \left(\frac{1}{t}\right)_{a}^{a-1} = 1$$

$$\frac{2}{3a(a-1)}=1$$

$$3a^2 - 3a - 2 = 0$$

$$a = \frac{3 \pm \sqrt{33}}{6}$$

10. Let a_1 , a_2 , a_3 , be an arithmetic progression with a_1 = 7 and common difference 8. Let T_1 , T_2 , T_3 ,... be such that $T_1 = 3$ and $T_{n+1} - T_n = a_n$ for $n \ge 1$. Then, which of the following is/are TRUE?

(B)
$$\sum_{k=1}^{20} T_k = 10510$$
 (C) $T_{30} = 3454$

(C)
$$T_{30} = 3454$$

(D)
$$\sum_{k=1}^{30} T_k = 35610$$

Ans. (B,C)

Sol.
$$a_1 = 7$$
, $d = 8$

$$T_{n+1} - T_n = a_n \forall n \ge 1$$

$$S_n = T_1 + T_2 + T_3 + \dots + T_{n-1} + T_n$$

on subtraction

$$T_n = T_1 + a_1 + a_2 + \dots + a_{n-1}$$

$$T_n = 3 + (n-1)(4n-1)$$

$$T_n = 4n^2 - 5n + 4$$

$$\sum_{k=1}^{n} T_{k} = 4 \sum n^{2} - 5 \sum n + 4n$$

$$T_{20} = 1504$$

$$T_{30} = 3454$$

$$\sum_{k=1}^{30} T_k = 35615$$

$$\sum_{k=1}^{20} T_k = 10510$$

11. Let P₁ and P₂ be two planes given by

$$P_1$$
: $10x + 15y + 12z - 60 = 0$

$$P_2$$
: $-2x + 5y + 4z - 20 = 0$.

Which of the following straight lines can be an edge of some tetrahedron whose two faces lie on P1 and

(A)
$$\frac{x-1}{0} = \frac{y-1}{0} = \frac{z-1}{5}$$
 (B) $\frac{x-6}{-5} = \frac{y}{2} = \frac{z}{3}$ (C) $\frac{x}{-2} = \frac{y-4}{5} = \frac{z}{4}$ (D) $\frac{x}{1} = \frac{y-4}{-2} = \frac{z}{3}$

(C)
$$\frac{x}{-2} = \frac{y-4}{5} = \frac{z}{4}$$

(D)
$$\frac{x}{1} = \frac{y-4}{-2} = \frac{z}{3}$$

(A,B,D)Ans.

Sol. line of intersection is
$$\frac{x}{0} = \frac{y-4}{-4} = \frac{z}{5}$$

- (1) Any skew line with the line of intersection of given planes can be edge of tetrahedron.
- (2) any intersecting line with line of intersection of given planes must lie either in plane P₁ or P₂ can be edge of tetrahedron.
- 12. Let S be the reflection of a point Q with respect to the plane given by

$$\vec{r} = -(t+p)\hat{i} + t\hat{i} + (1+p)\hat{k}$$

where t, p are real parameters and $\hat{i}, \hat{j}, \hat{k}$ are the unit vectors along the three positive coordinate axes. If the position vectors of Q and S are $10\hat{i} + 15\hat{j} + 20\hat{k}$ and $\alpha\hat{i} + \beta\hat{j} + \gamma\hat{k}$ respectively, then which of the following is/are TRUE?

(A)
$$3(\alpha + \beta) = -101$$

(B)
$$3(\beta + \gamma) = -71$$

$$(C) 3(\gamma + \alpha) = -86$$

(B)
$$3(\beta + \gamma) = -71$$
 (C) $3(\gamma + \alpha) = -86$ (D) $3(\alpha + \beta + \gamma) = -121$

Ans. (A,B,C)

Sol.
$$\vec{r} = \hat{k} + t(-\hat{i} + \hat{j}) + p(-\hat{i} + \hat{k})$$

$$\vec{n} = \hat{i} + \hat{j} + \hat{k}$$

$$\Rightarrow$$
 x + y + z = 1

Q(10,15,20) and S(α , β , γ)

$$\frac{\alpha - 10}{1} = \frac{\beta - 15}{1} = \frac{\gamma - 20}{1} = -2\left(\frac{10 + 15 + 20 - 1}{1 + 1 + 1}\right)$$

$$=-\frac{88}{3}$$

$$\Rightarrow \left(\alpha,\beta,\gamma\right) \equiv \left(-\frac{58}{3},-\frac{43}{3},-\frac{28}{3}\right)$$

Consider the parabola $y^2 = 4x$. Let S be the focus of the parabola. A pair of tangents drawn to the 13. parabola from the point P = (-2, 1) meet the parabola at P₁ and P₂. Let Q₁ and Q₂ be points on the lines SP₁ and SP₂ respectively such that PQ₁ is perpendicular to SP₁ and PQ₂ is perpendicular to SP₂. Then, which of the following is/are TRUE?

(A)
$$SQ_1 = 2$$

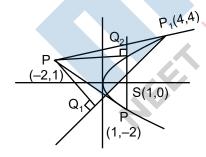
(B)
$$Q_1Q_2 = \frac{3\sqrt{10}}{5}$$

(C)
$$PQ_1 = 3$$

(D)
$$SQ_2 = 1$$

(B,C,D)Ans.

Let equation of tangent with slope 'm' be Sol.



$$T: y = mx + \frac{1}{m}$$

T: passes through (-2, 1) so

$$1 = -2m + \frac{1}{m}$$

$$\Rightarrow$$
 m = -1orm = $\frac{1}{2}$

Points are given by $\left(\frac{a}{m^2}, \frac{2a}{m}\right)$

So, one point will be (1, -2) & (4, 4)

Let $P_1(4, 4) \& P_2(1, -2)$

 $P_1S: 4x - 3y - 4 = 0$

 $P_2S: x-1=0$

$$PQ_1 = \left| \frac{4(-2) - 3(1) - 4}{5} \right| = 3$$

$$SP = \sqrt{10}; PQ_2 = 3; SQ_1 = 1 = SQ_2$$

$$\frac{1}{2} \left(\frac{Q_1 Q_2}{2} \right) \times \sqrt{10} = \frac{1}{2} \times 3 \times 1$$
 (comparing Areas)

$$\Rightarrow Q_1Q_2 = \frac{2 \times 3}{\sqrt{10}} = \frac{3\sqrt{10}}{5}$$

Let |M| denote the determinant of a square matrix M. Let $g: \left[0, \frac{\pi}{2}\right] \to \square$ be the function defined by 14.

$$g(\theta) = \sqrt{f(\theta) - 1} + \sqrt{f\bigg(\frac{\pi}{2} - \theta\bigg) - 1}$$

where

$$f(\theta) = \frac{1}{2} \begin{vmatrix} 1 & \sin\theta & 1 \\ -\sin\theta & 1 & \sin\theta \\ -1 & -\sin\theta & 1 \end{vmatrix} + \begin{vmatrix} \sin\pi & \cos\left(\theta + \frac{\pi}{4}\right) & \tan\left(\theta - \frac{\pi}{4}\right) \\ -\cos\frac{\pi}{2} & \log_e\left(\frac{4}{\pi}\right) \\ \cot\left(\theta + \frac{\pi}{4}\right) & \log_e\left(\frac{\pi}{4}\right) & \tan\pi \end{vmatrix}$$

Let p(x) be a quadratic polynomial whose roots are the maximum and minimum values of the function $g(\theta)$, and $p(2) = 2 - \sqrt{2}$. Then, which of the following is/are TRUE?

(A)
$$p\left(\frac{3+\sqrt{2}}{4}\right) < 0$$

(B)
$$p\left(\frac{1+3\sqrt{2}}{4}\right) > 0$$

(A)
$$p\left(\frac{3+\sqrt{2}}{4}\right) < 0$$
 (B) $p\left(\frac{1+3\sqrt{2}}{4}\right) > 0$ (C) $p\left(\frac{5\sqrt{2}-1}{4}\right) > 0$ (D) $p\left(\frac{5-\sqrt{2}}{4}\right) < 0$

(D)
$$p\left(\frac{5-\sqrt{2}}{4}\right) < 0$$

(A, C) Ans.

$$\begin{aligned} \textbf{Sol.} \qquad f(\theta) = \frac{1}{2} \begin{vmatrix} 1 & \sin\theta & 1 \\ -\sin\theta & 1 & \sin\theta \\ -1 & -\sin\theta & 1 \end{vmatrix} + \begin{vmatrix} \sin\pi & \cos\left(\theta + \frac{\pi}{4}\right) & \tan\left(\theta - \frac{\pi}{4}\right) \\ \sin\left(\theta - \frac{\pi}{4}\right) & -\cos\frac{\pi}{2} & \log_{e}\left(\frac{4}{\pi}\right) \\ \cot\left(\theta + \frac{\pi}{4}\right) & \log_{e}\frac{\pi}{4} & \tan\pi \end{vmatrix} \end{aligned}$$

$$f(\theta) = \frac{1}{2} \begin{vmatrix} 2 & \sin\theta & 1 \\ 0 & 1 & \sin\theta \\ 0 & -\sin\theta & 1 \end{vmatrix} + \begin{vmatrix} 0 & -\sin\left(\theta - \frac{\pi}{4}\right) & \tan\left(\theta - \frac{\pi}{4}\right) \\ \sin\left(\theta - \frac{\pi}{4}\right) & 0 & \log_e\left(\frac{4}{\pi}\right) \\ -\tan\left(\theta - \frac{\pi}{4}\right) & -\log_e\left(\frac{4}{\pi}\right) & 0 \end{vmatrix}$$

 $f(\theta) = (1 + \sin^2 \theta) + 0$ (skew symmetric)

$$g(\theta) = \sqrt{f(\theta) - 1} + \sqrt{f\bigg(\frac{\pi}{2} - \theta\bigg) - 1}$$

$$= \mid \sin\theta \mid + \mid \cos\theta \mid \qquad \text{for } \theta \in \left[0, \frac{\pi}{2}\right]$$

$$g(\theta) \in [1, \sqrt{2}]$$

Again let $P(x) = k(x - \sqrt{2})(x - 1)$

$$2-\sqrt{2}=k(2-\sqrt{2})(2-1)$$

$$\Rightarrow$$
 k = 1 (P(2) = 2 - $\sqrt{2}$ again)

$$\therefore P(x) = (x - \sqrt{2})(x - 1)$$

...ect $4^{-\frac{1}{2}} < 0 \text{ incorrect}$ $\text{pulon (C) } P\left(\frac{5\sqrt{2}-1}{4}\right) > 0 \text{ correct}$ $\text{option (D) } P\left(\frac{5-\sqrt{2}}{4}\right) > 0 \text{ incorrect}$

option (B)
$$P\left(\frac{1+3\sqrt{2}}{4}\right) < 0$$
 incorrect

option (C)
$$P\left(\frac{5\sqrt{2}-1}{4}\right) > 0$$
 correct

option (D)
$$P\left(\frac{5-\sqrt{2}}{4}\right) > 0$$
 incorrect

SECTION-3: (Maximum Marks: 12)

- This section contains FOUR (04) Matching List Sets.
- Each set has **ONE** Multiple Choice Question.
- Each set has TWO lists : List-I and List-II.
- List-I has Four entries (I), (II), (III) and (IV) and List-II has Five entries (P), (Q), (R), (S) and (T).
- **FOUR** options are given in each Multiple Choice Question based on **List-I** and **List-II** and **ONLY ONE** of these four options satisfies the condition asked in the Multiple Choice Question.
- Answer to each question will be evaluated according to the following marking scheme:

Full Marks : +3 **ONLY** if the option corresponding to the correct combination is chosen;

Zero Marks : 0 If none of the options is chosen (i.e. the question is unanswered);

Negative Marks : -1 In all other cases.

15. Consider the following lists:

List I

(I)
$$\left\{x \in \left[-\frac{2\pi}{3}, \frac{2\pi}{3}\right] : \cos x + \sin x = 1\right\}$$

(II)
$$\left\{x \in \left[-\frac{5\pi}{18}, \frac{5\pi}{18} \right] : \sqrt{3} \tan 3x = 1 \right\}$$

(III)
$$\left\{x \in \left[-\frac{6\pi}{5}, \frac{6\pi}{5}\right] : 2\cos(2x) = \sqrt{3}\right\}$$

(IV)
$$\left\{ x \in \left[-\frac{7\pi}{4}, \frac{7\pi}{4} \right] : \sin x - \cos x = 1 \right\}$$

The correct option is:

Ans. (B)

Sol. (I)
$$\left\{x \in \left[\frac{-2\pi}{3}, \frac{2\pi}{3}\right] : \cos x + \sin x = 1\right\}$$

cosx + sinx = 1

$$\Rightarrow \frac{1}{\sqrt{2}}\cos x + \frac{1}{\sqrt{2}}\sin x = \frac{1}{\sqrt{2}}$$

$$\Rightarrow \cos\left(x - \frac{\pi}{4}\right) = \cos\frac{\pi}{4}$$

$$\Rightarrow x - \frac{\pi}{4} = 2n\pi \pm \frac{\pi}{4}; n \in Z$$

$$\Rightarrow x=2n\pi; x=2n\pi+\frac{\pi}{2}; n\in Z$$

List II

- (P) has two elements
- (Q) has three elements
- (R) has four elements
- (S) has five elements
- (T) has six elements

 \Rightarrow x $\in \left\{0, \frac{\pi}{2}\right\}$ in given range has two solutions

(II)
$$\left\{x \in \left[\frac{-5\pi}{18}, \frac{5\pi}{18}\right] : \sqrt{3} \tan 3x = 1\right\}$$

$$\sqrt{3} \tan 3x = 1 \Rightarrow \tan 3x = \frac{1}{\sqrt{3}} \Rightarrow 3x = n\pi + \frac{\pi}{6}$$

$$\Rightarrow x = (6n+1)\frac{\pi}{18}; n \in Z$$

 \Rightarrow x $\in \left\{ \frac{\pi}{18}, \frac{-5\pi}{18} \right\}$ in given range has two solutions

(III)
$$\left\{x \in \left[-\frac{6\pi}{5}, \frac{6\pi}{5}\right] : 2\cos(2x) = \sqrt{3}\right\}$$

$$2\cos 2x = \sqrt{3}$$

$$\Rightarrow \cos 2x = \frac{\sqrt{3}}{2} = \cos \frac{\pi}{6}$$

$$\Rightarrow 2x = 2n\pi \pm \frac{\pi}{6}; n \in Z$$

$$\Rightarrow x = n\pi \pm \frac{\pi}{12}; n \in Z$$

$$X \in \left\{\pm \frac{\pi}{12}, \pi \pm \frac{\pi}{12}, -\pi \pm \frac{\pi}{12}\right\}$$

Six solutions in given range

(Iv)
$$\left\{x \in \left[-\frac{7\pi}{4}, \frac{7\pi}{4}\right] : \sin x - \cos x = 1\right\}$$

$$\cos x - \sin x = -1$$

$$(\pi) -1 3\pi$$

$$\cos x - \sin x = -1$$

$$\Rightarrow \cos\left(x + \frac{\pi}{4}\right) = \frac{-1}{\sqrt{2}} = \cos\frac{3\pi}{4}$$

$$\Rightarrow x + \frac{\pi}{4} = 2n\pi \ge \frac{3\pi}{4}; n \in Z$$

$$\Rightarrow x = 2n\pi + \frac{\pi}{2} \text{ or } x = 2n\pi - \pi; n \in Z$$

$$\Rightarrow$$
 x \in $\left\{\frac{\pi}{2}, \frac{-3\pi}{2}, \pi, -\pi\right\}$ four solutions in given range

16. Two players, P₁ and P₂, play a game against each other. In every round of the game, each player rolls a fair die once, where the six faces of the die have six distinct numbers. Let x and y denote the readings on the die rolled by P_1 and P_2 , respectively. If x > y, then P_1 scores 5 points and P_2 scores 0 point. If x = y, then each player scores 2 points. If x < y, then P_1 scores 0 point and P_2 scores 5 points. Let X_i and Y_i be the total scores of P_1 and P_2 , respectively, after playing the i^{th} round.

List II

List I

(I) Probability of $(X_2 \ge Y_2)$ is (P) $\frac{3}{8}$

(II) Probability of $(X_2 \ge Y_2)$ is (Q) $\frac{11}{16}$

(III) Probability of $(X_2 \ge Y_2)$ is (R) $\frac{5}{16}$

(IV) Probability of $(X_2 \ge Y_2)$ is $(S) \frac{355}{864}$

(T) $\frac{77}{432}$

The correct option is:

(A) (I) - (Q); (II) - (R); (III) - (T); (IV) - (S) (B) (I) - (Q); (II) - (R); (III) - (T); (IV) - (T)

Ans. (A)

Sol. P(draw in 1 round) = $\frac{6}{36} = \frac{1}{6}$

P(draw in 1 round) = $\frac{1}{2} \left(1 - \frac{1}{6} \right) = \frac{5}{12}$

P(draw in 1 round) = $\frac{5}{12}$

 $P(X_2 > Y_2) = P(10,0) + P(7,2) = \frac{5}{12} \times \frac{5}{12} + \frac{5}{12} \times \frac{1}{6} \times 2 = \frac{45}{144} = \frac{5}{16}$

 $P(X_2 = Y_2) = P(5,5) + P(4,4) = \frac{5}{12} \times \frac{5}{12} \times 2 + \frac{1}{6} \times \frac{1}{6} = \frac{25+2}{72} = \frac{3}{8}$

 $P(X_3 = Y_3) = P(6,6) + P(7,7) = \frac{1}{6 \times 6 \times 6} + \frac{5}{12} \times \frac{1}{6} \times \frac{5}{12} \times 6 = \frac{2}{432} + \frac{75}{432} = \frac{77}{432}$

 $P(X_3 > Y_3) = \frac{1}{2} \left(1 - \frac{77}{432}\right) = \frac{355}{864}$

17. Let p, q, r be nonzero real numbers that are, respectively, the 10th, 100th and 1000th terms of a harmonic progression. Consider the system of linear equations

$$x + y + z = 1$$

$$10x + 100y + 1000z = 0$$

$$qr x + pr y + pq z = 0$$
.

List I

(I) If $\frac{q}{r} = 10$, then the system of linear equations has

(P) $x = 0, y = \frac{10}{9}, z = -\frac{1}{9}$ as a solution

(II) If $\frac{p}{r} \neq 100$, then the system of linear equations has

(Q) $x = \frac{10}{9}, y = -\frac{1}{9}, z = 0$ as a solution

(III) If $\frac{p}{q} \neq 10$, then the system of linear equations has

(R) infinitely many solution

(IV) If $\frac{q}{a} = 10$, then the system of linear equations has

(S) no solution

List II

(T) at least one solution

The correct option is:

(A) (I) - (T); (II) - (R); (III) - (S); (IV) - (T)

(B) (I) - (Q); (II) - (S); (III) - (S); (IV) - (R)

(C) (I) - (Q); (II) - (R); (III) - (P); (IV) - (R)

(D) (I) - (T); (II) - (S); (III) - (P); (IV) - (T) FOUND WILL

Ans.

If $\frac{q}{r} = 10 \Rightarrow A = D \Rightarrow D_x = D_y = D_z = 0$ Sol.

So, there are infinitely many solutions

Look of infinitely many solutions can be given as

$$x + y + z = 1$$

&
$$10x + 100y + 1000z = 0 \Rightarrow x + 10y + 100z = 0$$

Let $z = \lambda$

then $x + y = 1 - \lambda$

and $x + 10y = -100\lambda$

$$\Rightarrow x = \frac{10}{9} + 10\lambda; y = \frac{-1}{9} - 11\lambda$$

Let
$$z = \lambda$$

then $x + y = 1 - \lambda$
and $x + 10y = -100\lambda$

$$\Rightarrow x = \frac{10}{9} + 10\lambda; y = \frac{-1}{9} - 11\lambda$$
i.e., $(x,y,z) = \left(\frac{10}{9} + 10\lambda, \frac{-1}{9} - 11\lambda, \lambda\right)$

$$Q\bigg(\frac{10}{9},\frac{-1}{9},0\bigg) \text{valid for } \lambda=0$$

$$P\left(0,\frac{10}{9},\frac{-1}{9}\right)$$
 not valid for any λ

$$(I) \rightarrow Q,R,T$$

(II) If
$$\frac{p}{r} \neq 100$$
, then Dy $\neq 0$

So no solution

$$(II) \rightarrow (S)$$

- (III) If $\frac{p}{q} \neq 10$, then $D_z \neq 0$ so, no solution
- $(III) \rightarrow (S)$
- (IV) If $\frac{p}{q} = 10 \Rightarrow D_z = 0 \Rightarrow D_x = D_y = 0$

So infinitely many solution

- $(IV) \rightarrow Q,R,T$
- 18. Consider the ellipse

$$\frac{x^2}{4} + \frac{y^2}{3} = 1.$$

Let $H(\alpha, 0)$, $0 < \alpha < 2$, be a point. A straight line drawn through H parallel to the y-axis crosses the ellipse and its auxiliary circle at points E and F respectively, in the first quadrant. The tangent to the ellipse at the point E intersects the positive x-axis at a point G. Suppose the straight line joining F and the origin makes an angle ϕ with the positive x-axis.

List II

List I

- (I) If $\phi = \frac{\pi}{4}$, then the area of the triangle FGH is (P) $\frac{(\sqrt{3}-1)^2}{8}$
- (II) If $\phi = \frac{\pi}{3}$, then the area of the triangle FGH is (Q) 1
- (III) If $\phi = \frac{\pi}{6}$, then the area of the triangle FGH is (R)
- (IV) If $\phi = \frac{\pi}{12}$, then the area of the triangle FGH is (S) $\frac{1}{2\sqrt{3}}$
 - $(T) \ \frac{3\sqrt{3}}{2}$

The correct option is:

- (A) (I) (R); (II) (S); (III) (Q); (IV) (P)
- (B) (I) (R); (II) (T); (III) (S); (IV) (P)
- (C) (I) (Q); (II) (T); (III) (S); (IV) (P)
- (D) (I) (Q); (II) (S); (III) (Q); (IV) (P)

Ans. (C

- Sol. Let F(2cos\u00f3, 2sin\u00f4)
 - & E(2cos ϕ , $\sqrt{3}$ sin ϕ)

$$EG: \frac{x}{2}\cos\varphi + \frac{y}{\sqrt{3}}\sin\varphi = 1$$

$$\therefore G\left(\frac{2}{\cos\phi},0\right) \text{ and } \alpha = 2\cos\phi$$

$$ar(\Delta FGH) = \frac{1}{2}HG \times FH$$

$$=\frac{1}{2}\left(\frac{2}{\cos\phi}-2\cos\phi\right)\times2\sin\phi$$

$$f(\phi) = 2\tan\phi\sin^2\phi$$

$$\therefore \text{ (I) } f\left(\frac{\pi}{4}\right) = 1 \text{ (II) } f\left(\frac{\pi}{3}\right) = \frac{3\sqrt{3}}{2} \text{ (III) } f\left(\frac{\pi}{6}\right) = \frac{1}{2\sqrt{3}}$$

$$(IV) f\left(\frac{\pi}{12}\right) = 2(2 - \sqrt{3}) \left(\frac{\sqrt{3} - 1}{2\sqrt{2}}\right)^2 = (4 - 2\sqrt{3}) \frac{(\sqrt{3} - 1)^2}{8} = \frac{(\sqrt{3} - 1)^4}{8}$$

$$\therefore (I) \to (Q) ; (II) \to (T) ; (III) \to (S) ; (IV) \to (P)$$

$$\therefore \ (I) \rightarrow (Q) \ ; \ (II) \rightarrow (T) \ ; \ (III) \rightarrow (S) \ ; \ (IV) \rightarrow (P)$$

